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based on the assumption that atoms tend to have either empty valence shells or full valence shells of eight electrons. (p. 159) ion.

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____ 5. Mendeleev is credited with developing the first

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successful. a. periodic table. b. method for determining atomic number. c. test for radioactivity. d. use of X rays. ____ 6.

Mendeleev did not always list elements in his periodic table in order of increasing atomic mass because he grouped together elements with similar. a. properties. c ...

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Modern Chemistry 34 Chapter Test Name Class Date Chapter Test B, continued PART IV Write the answers to the following

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questions in the space provided. 27. What two early twentieth-century observations involving the interaction of light and matter could not be explained solely by the wave theory of light? 28.

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