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actinides. a special block of metallic

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elements with atomic numbers 89-103.
alkali metals. group 1 elements, except
for hydrogen.

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emrosepow. Terms in this set (33)
Periodic Table. An arrangement of elements in columns, based on a set of properties that repeat from row to row.
Period. A row in a periodic table of elements. Group.

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Chapter 5 Periodic Table

Section 5.1 – Organizing the Elements A periodic table is an arrangement of elements in columns, based on a set of properties that repeat from row to row. Dmitri Mendeleev is credited with creating the first useful periodic table.

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B. Use the Periodic Table to predict the above properties. Section 5.1 – Organizing the Elements A periodic table is an arrangement of elements in columns, based on a set of properties that repeat from row to row.

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Chapter 5: periodic table. periodic law. atomic radius. ion. ionization energy. the physical and chemical properties of the elements are perio.... one-half the distance between the nuclei of identical atoms th.... an atom or group of bonded atoms that has a positive or negati....

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Chapter 5 - The Periodic Law 5-1 History of the Periodic Table A. - Mendeleev and Chemical Periodicity 1. Dmitri Mendeleev went about organizing the elements according to their _____ as you might organize information for a _____.

He placed the name of each known element on a _____, together with the

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atomic _____ of the element and ...

Chapter 5 - The Periodic Law

Chapter 5: The Periodic Table (Test Answers) STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. chxistina. Terms in this set (36) The order of elements in the periodic table is based on atomic mass

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the number of protons in the nucleus the
number of neutrons in the nucleus

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Periodic Classification of Elements, Class 10 Chapter 5 Science Notes includes detailed explanations about the modern periodic table, periodic trends and about

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is discovery.

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Chapter 5. Periodic Law. 2. Ionization Energy
Electrons can be removed from an atom if enough energy is supplied
Using A as a symbol for an atom of ANY element, the process can be expressed

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as follows: $A + \text{energy} \rightarrow A^+ + e^-$ A^+ represents an ion of element A with single positive charge (a 1+ ion) Ion an atom or group of bonded atoms that have a positive or negative charge Ionization any process that results in the formation of an ion Period Trends In general, ionization energies of the ...

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5.2 Objectives . Explain the relationship between electrons in sublevels and the length of each period of the periodic table.; Locate and name the four blocks

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of the periodic table. Explain the reasons for these names. Discuss the relationship between group configurations and group numbers.; Describe the locations in the periodic table and the general properties of the alkali metals, the ...

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Chapter 5 - Periodic Classification of Elements In standard 9 we have learned that matter around us is present in the form of elements, compounds and mixtures and the elements contain atoms of ...

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Elements

The Periodic Table of the Elements is useful for revealing patterns and trends in the elements. Which statement accurately describes a pattern in the size of atomic radii in the Periodic Table of the Elements?

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